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web log 10^{-4} 0.5891 $\times 10^{-5}$ mol dm^{-3} log pK_a 3.75 4.05 0.30 10.0 30.0 50.0 and HCOOH 0.1 so HCOO^- 0.200 mol dm^{-3} amount $\text{vol} \times$

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