



6

(b) $\text{Cl}_2(\text{g})$ dissolves in cold water and reacts with it.



(i) Identify the oxidation number of chlorine in each of the chlorine-containing species in this reaction.

chlorine-containing species	Cl_2	$\text{HCl}^{\text{+1}}$	$\text{HClO}^{\text{+1} \text{+2}}$
oxidation number of chlorine	0	-1	+1

[2]

(ii) Name the type of reaction occurring.

solved by TheamKL

Disproportionation

[1]

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Lauren Gardner



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