Hydrocarbons

Methane	General formula for an alkane		General formula for an alkene	Ethene	Propene	Butene	
	How to test between an alkene and alkane		Alkane or alkene? C/H ₄ C/H ₆ C/H ₆ C/H ₆ C/H ₆				
Ethane				Show how ethere is polymerised			
	Definition of a hydrocarbon						
Propune		What is meant by saturated/unsaturated		Draw the monomer that makes this polymer Br H Br H Br H I I I I I H-C-C-C-C-C-C-H I I I I I H H H H H H			
Butane		How ethanol is produced from ethane		Describe the proce Process:	ns of cracking	Complete the equations for cracking C ₀ H ₁₀ → C ₁ H ₄ + + C ₄ H ₃₈	
A way to remember the order of alkanes		_by fermentation		Types of products. Conditions: Why it is done:	The types of products produced:		

Hydrocarbon Compounds Core Teaching Resources

Jay K. Hackett

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discuss all science and technology aspects that are relevant to sustainable human quality development We also expect that the future ICoSTA conference will be as successful and stimulating as indicated by the contributions presented in this volume

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