HOLT MODERN CHEMISTRY -- CHAPTER 1 REVIEW

The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

***** Test items will be recall, examples, and/or application of this content. *****

OUTCOMES

- 1.1: Distinguish between types of research basic, applied, and technological development. (F & PK)
- 1.2: Distinguish between the physical and chemical properties of matter. (F & PK)
- 1.2: Distinguish among homogeneous and heterogeneous mixtures, elements, and compounds. (F & PK)
- 1.3: List properties of metals and non-metals. (T & R)
- 2.3: Determine significant figures. (F & PK)

1.1: CHEMISTRY IS A PHYSICAL SCIENCE

Vocabulary

- chemistry the scientific study of the composition, structure, and properties of matter and the changes that matter undergoes
- chemical -- any substance with a defined composition

Chapter Highlights

- Chemistry is the study of the composition, structure, and properties of matter and the changes that matter undergoes.
- A chemical is any substance that has a definite composition or is used or produced in a chemical process.
- Basic research is carried out for the sake of increasing knowledge. Applied research is carried out to solve practical problems. Technological development involves the use of existing knowledge to make life easier or more convenient.

1.2: MATTER AND ITS PROPERTIES

Vocabulary

- matter anything that has mass and takes up space
- mass a measure of the amount of matter in an object
- element a substance that cannot be separated or broken down into simpler substances by chemical means; all atoms of an element have the same atomic number
- atom -- the smallest unit of an element that maintains the chemical properties of that element.
- compound -- a substance made up of atoms of two or more different elements joined by chemical bonds
- extensive property -- a property that depends on the extent or size of a system.
- intensive property a property that does not depend on the amount of matter present, such as pressure, temperature, or density
- physical property a characteristic of a substance that does not involve a chemical change, such as density, color, or hardness
- physical change -- a change of matter from one form to another without a change in chemical properties
- change of state -- the change of a substance from one physical state to another
- solid -- the state of matter in which the volume and shape of a substance are fixed
- liquid -- the state of matter that has a definite volume but not a definite shape.
- qas -- a form of matter that does not have a definite volume or shape
- plasma -- in physical science, a state of matter that starts as a gas and then becomes ionized; it consists of freemoving ions and electrons and it takes on an electric charge
- chemical property a property of matter that describes a substance's ability to participate in chemical reactions
- chemical change -- a change that occurs when one or more substances change into entirely new substances with different properties
- chemical reaction the process by which one or more substances change to produce one or more different substances
- reactant -- a substance or molecule that participates in a chemical reaction
- product -- a substance that forms in a chemical reaction

Holt Mcdougal Modern Chemistry Chapter 1review Answers

Samuel Madras, McGraw-Hill Company of Canada

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Modern Biology Towle, Albert Towle,1991 **Lab Experiments Modern Chemistry** Harold Clark Metcalfe, 1982 Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows guick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Objective Tests for Basic Modern Chemistry: Part One, Chapters 1 to 21 Samuel Madras, McGraw-Hill Company of Canada, 1964 **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 The Comprehensive General and Organic Chemistry Review Guide - Volume 2 Leonard Powell D 0,2020-09-03 This guide begins with a survey of topics encountered in a General Chemistry 1 and General Chemistry 2 course Chapters 1 15 Organic chemistry is introduced in Chapter 16 the topics of Chapters 16 31 are typically taught in Organic Chemistry 1 and 2 courses Chapter 32 is an optional advanced topics chapter to entice those interested in further study of either pure physical or applied synthetic organic chemistry Volumes 1 and 2 will contain theory and concepts with examples within the text Volume 3 will contain additional worked examples as well as problems for practice Holt Mcdougal Modern Chemistry Texas Holt Chemistry Holt, Rinehart and Winston Staff, Holt, Rinehart, and Winston, inc, 1996-01-01 Hart Mcdougal, 2014-01-01 Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2006-01-01

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