

Making Salts From Metal Carbonates

Adding a metal carbonate or a metal hydrogen carbonate to an acid produces a salt, carbon dioxide and water.

acid	+	metal carbonate	➡	a salt	+	carbon dioxide	water
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hydrochloric acid	+	calcium carbonate	➡	calcium chloride	+	carbon dioxide	water
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hydrochloric acid	+	sodium hydrogen carbonate	➡	sodium chloride	+	carbon dioxide	water
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Metal carbonates contain a carbonate ion (CO_3^{2-}) and therefore this reaction with acid is often used as a chemical test for this ion.

Metal Carbonates Isa

PT Brinkman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the PT Brinkman text.

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