

Chapter 8 Chemical Reactions Chapter Review Answers

1. The evolution of energy, the production of a gas, the formation of a precipitate, and a color change.
2. A correctly written chemical equation must represent the known facts, contain the correct formulas for the reactants and products, and satisfy the law of conservation of mass.
3.
 - a. It is a number that appears in front of a formula in a chemical equation.
 - b. The number of atoms of each type in the formula is multiplied by the coefficient.
4. methane + oxygen $\rightarrow$ carbon dioxide + water;
 $\text{CH}_4(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g})$;
 $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$
5. A chemical equation provides the number of molecules, moles, and atoms of each element or compound composing the reactants and products as well as the mole ratios of the reactants to each other and to the products.
6. Word and formula equations have only qualitative meaning; they do not express the relative quantities of a reaction's reactants or products.
7.
 - a. a solution in water
 - b. a substance that accelerates a chemical reaction but can be recovered unchanged
 - c. a chemical reaction in which the products re-form the original reactants
8.
 - a. KOH
 - b. $\text{Ca}(\text{NO}_3)_2$
 - c. Na_2CO_3
 - d. CCl_4
 - e. MgBr_2
9. (1) Balance different types of atoms one at a time; (2) balance types of atoms that appear only once on each side of the equation first; (3) balance as single units any polyatomic ions that appear on both sides of the equation; and (4) balance H atoms and O atoms last.
10.
 - a. 6N
 - b. 4H, 2O
 - c. 4H, 4N, 12O
 - d. 2Ca, 4O, 4H
 - e. 3Ba, 6Cl, 18O
 - f. 5Fe, 10N, 30O
 - g. 12Mg, 8P, 32O
 - h. 4N, 16H, 2S, 8O
 - i. 12Al, 188Se, 72O
 - j. 12C, 32H
11.
 - a. $2\text{ZnS}(\text{s}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{ZnO}(\text{s}) + 2\text{SO}_2(\text{g})$
 - b. $2\text{HCl}(\text{aq}) + \text{Ba}(\text{OH})_2(\text{aq}) \rightarrow \text{BaCl}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$
 - c. $2\text{HNO}_3(\text{aq}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}(\text{NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$
12.
 - a. Solid zinc sulfide reacts with oxygen gas to form solid zinc oxide and sulfur dioxide gas.
 - b. When solid calcium hydride is added to water, aqueous calcium hydroxide and hydrogen gas are formed.
 - c. Aqueous silver nitrate mixed with aqueous potassium iodide produces a precipitate of silver iodide and aqueous potassium nitrate.
13.
 - a. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - b. $2\text{Al} + \text{Fe}_2\text{O}_3 \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$
 - c. $\text{Pb}(\text{CH}_3\text{COO})_2 + \text{H}_2\text{S} \rightarrow \text{PbS} + 2\text{CH}_3\text{COOH}$
14.
 - a. LiO_2 is an incorrect formula; $4\text{Li} + \text{O}_2 \rightarrow 2\text{Li}_2\text{O}$
 - b. H_2Cl_2 is an incorrect formula; $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - c. MgO_2 is an incorrect formula, and the equation as written is not balanced; $\text{MgCO}_3 \rightarrow \text{MgO} + \text{CO}_2$
 - d. I is an incorrect formula for iodine, and the equation is not balanced; $2\text{NaI} + \text{Cl}_2 \rightarrow 2\text{NaCl} + \text{I}_2$
15.
 - a. $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$
 - b. $\text{P}_2\text{O}_{10} + 6\text{H}_2\text{O} \rightarrow 4\text{H}_3\text{PO}_4$
 - c. $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$
16. $\text{CH}_4(\text{g}) + 4\text{Cl}_2(\text{g}) \rightarrow \text{CCl}_4(\text{l}) + 4\text{HCl}(\text{g})$
17.
 - a. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
 - b. $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
 - c. $2\text{Li} + \text{Cl}_2 \rightarrow 2\text{LiCl}$
 - d. $\text{Ca} + \text{I}_2 \rightarrow \text{CaI}_2$
18.
 - (1) In a synthesis reaction, two or more substances combine to form a new substance: $\text{A} + \text{X} \rightarrow \text{AX}$
 - (2) In a decomposition reaction, a single compound undergoes a reaction that produces two or more simpler substances: $\text{AX} \rightarrow \text{A} + \text{X}$
 - (3) In a single-displacement reaction, one element replaces another element in a compound:
 $\text{A} + \text{BX} \rightarrow \text{AX} + \text{B}$ or $\text{Y} + \text{RX} \rightarrow \text{RY} + \text{X}$

Merrill Chemistry Ch 8 Review Answers

Sabine Zange



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Merrill Earth Science* Ralph M. Feather, Susan Leach Snyder, Dale T. Hesser, 1995 Physical Chemistry of Metallurgical Processes, Second Edition Mohammad Shamsuddin, 2021-06-22 This updated second edition retains its classroom tested treatment of physical chemistry of metallurgical topics such as roasting of sulfide minerals matte smelting converting structure properties and theories of slag reduction of oxides and reduction smelting interfacial phenomena steelmaking secondary steelmaking role of halides in extraction of metals refining hydrometallurgy and electrometallurgy and adds new data in worked out examples as well as up to date references to the literature The book further explains the physical chemistry of various metallurgical topics steps involved in extraction of metals such as roasting matte smelting converting reduction smelting steelmaking reactions deoxidation stainless steelmaking vacuum degassing refining leaching chemical precipitation ion exchange solvent extraction cementation gaseous reduction and electrowinning Each topic is illustrated with appropriate examples of applications of the technique in extraction of some common reactive rare or refractory metal together with worked out problems explaining the principle of the operation The problems require imagination and critical analyses and also encourage readers for creative application of thermodynamic data in metal extraction Updates and condenses text throughout the book by sequential arrangement of paragraphs in different chapters Maximizes readers understanding of the physicochemical principles involved in extraction production of common and rare reactive metals by pyro as well as hydrometallurgical routes Reinforces concepts presented with worked examples in each chapter explaining the process steps Explains the physical chemistry of various metallurgical steps such as roasting matte smelting converting and reduction smelting steelmaking aqueous processing etc in extraction of metals Collects and uniformly presents scattered information on physicochemical principles of metal production from various books and journals Current Catalog National Library of Medicine (U.S.), First multi year cumulation covers six years 1965 70 **Energy Research Abstracts** ,1979 **Journal of Industrial and Engineering Chemistry** ,1916 **The Journal of Industrial and Engineering Chemistry** ,1918 *New Technical Books* New York Public Library, 1925 **Mining and Metallurgy** ,1925 The Bookseller, Newsdealer and Stationer ,1907 **Health Effects of Indoor Air Pollution** Mohammad Hadi Dehghani, Rama Rao Karri, Teresa Vera, Salwa Kamal Mohamed Hassan, 2024-04-01 Health Effects of Indoor Air Pollution Volume Two Air Pollution Human Health and the Environment is part of a three volume series This volume covers the various classifications of indoor air pollutants and discusses the health impact of indoor pollutants such as gaseous pollutants and particulate matter It also examines epidemiological studies related to different air pollutants on health and the workplace This book begins with an overview of classifications sources and occurrences of indoor air pollutants It also examines the environmental and health impacts due to organic and inorganic air pollutants and how to mitigate them through exposure and risk management Other sections explore

sick building syndrome which causes acute health and discomfort that appears to be linked to time spent in a building Recent trends and control strategies for occupation exposure due to poor indoor air quality in industrial and nonindustrial workplaces to human health are also covered This book is a valuable reference for academicians researchers and students in environmental health public health and occupational health as well as environmental engineers meteorologists epidemiologists medical researchers and environmental toxicologists Reviews sources and occurrences of organic and inorganic air pollutants Examines the health impact such as sick building syndrome and the effect of gaseous indoor pollutants and fine particulate matter on sensitive populations Discusses indoor air quality and the link between ventilation indoor air pollutants and environmental quality Resources in education ,1984-08 **Snack Foods** Sergio O. Serna-Saldivar,2022-04-20 The diverse segments of the snack industries that generate close to 520 billion of annual sales are adapting to new consumer s expectations especially in terms of convinience flavor shelf life and nutritional and health claims Snack Foods Processing Innovation and Nutritional Aspects was conceptualized to thoroughly cover practical and scientific aspects related to the chemistry technology processing functionality quality control analysis and nutrition and health implications of the wide array of snacks derived from grains fruits vegetables milk and meat poultry seafood This book focuses on novel topics influencing food product development like innovation new emerging technologies and the manufacturing of nutritious and health promoting snacks with a high processing efficiency The up to date chapters provide technical reviews emphasising flavored salty snacks commonly used as finger foods including popcorn wheat based products crispbreads pretzels crackers lime cooked maize snacks tortilla chips and corn chips extruded items expanded and half products or pellets potato chips peanuts almonds tree nuts and products derived from fruits vegetables milk animal and marine sources Key Features Describes traditional and novel processes and unit operatios used for the industrial production of plant and animal based snacks Depicts major processes employed for the industrial production of raw materials oils flavorings and packaging materials used in snack food operations Contains relevant and updated information about quality control and nutritional attributes and health implications of snack foods Includes simple to understand flowcharts relevant information in tables and recent innovations and trends Divided into four sections Snack Foods aims to understand the role of the major unit operations used to process snacks like thermal processes including deep fat frying seasoning packaging and the emerging 3 D printing technology Moreover the book covers the processing and characteristics of the most relevant raw materials used in snack operations like cereal based refined grits starches and flours followed by chapters for oils seasoning formulations and packaging materials The third and most extensive part of the book is comprised of several chapters which describe the manufacturing and quality control of snacks mentioned above The fourth section is comprised of two chapters related to the nutritional and nutraceutical and health promoting properties of all classes of snacks discussed herein

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hydrodynamics environment Nan Wu ,Ningning Zhang,Yan Liu,Fei Han,Cheng Zhang,Xingsen Guo,2024-01-29

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