

Solution
CHAPTER TEST
Class 11 - Chemistry

1. (a) Propene

Explanation: Elimination reaction takes place, a molecule of HBr gets eliminated to form propene.

2. (c) $\text{HI} > \text{HBr} > \text{HCl}$

Explanation: $\text{HI} > \text{HBr} > \text{HCl}$

3. (a) 1,3-dimethylbenzene



4. (d) Geometric isomers

Explanation: Isomers which have the same structural formulae but differ in the relative spatial arrangements of atoms or groups around the double bond are called geometric isomers. Thus geometrical isomers are a type of stereoisomers.

5. (d) HgSO_4

Explanation: Alkynes are not readily hydrated due to their low reactivity towards electrophilic addition reactions. Further dilute H_2SO_4 has no effect on carbon carbon triple bond. In the presence of HgSO_4 as catalyst, hydration occurs readily.

6. (a) A mixture of low boiling hydrocarbons.

Explanation: Pyrolysis is the decomposition of a compound by the application of heat. It involves the breaking of C-C and C-H bonds resulting in the formation of lower boiling alkanes.

7. (a) 3

Explanation: The position isomers of C_5H_8 are n-pentyne, 3-methyl butyne, pent-2-yne.

8. (a) C_8H_{16}

Explanation: Alkenes are unsaturated hydrocarbons with at least one double bond between the carbon atoms. Their general formula is C_nH_{2n} .

9. (b) Sodium acetate

Explanation:

This is an example of Kolbe's electrolysis method. The reaction is:



The step-wise redox reactions occurring in the electrolytic cell are depicted as under



At Anode: (Oxidation)



At Cathode: (Reduction)



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