

## Chapter Test A

### Chapter: Chemical Equations and Reactions

on the space provided below the letter of the item or phrase that best completes each statement or best answers each question.

1. The mass number of an atom is the sum of the number of \_\_\_\_\_ in the nucleus.
  - a. The number of protons and neutrons
  - b. The number of protons and electrons
  - c. The number of electrons in the outer shell and protons
  - d. None of the above
2. An element described by chemical equation:
 
$$\text{molecule} + \text{molecule} \rightarrow \text{molecule} + \text{molecule} + \text{molecule}$$
  - a. molecule is a product
  - b. molecule is a product
  - c. molecule is a reactant
  - d. None of the above
3. If the chemical equation is balanced, the number of \_\_\_\_\_ is the same on both sides.
  - a. The number of atoms of each element in the reaction (e.g. number of protons) of the elements
  - b. All of the reactants and products
  - c. The number of molecules of each element in the reaction (e.g. number of molecules of each element)
  - d. The number of molecules of each element in the reaction (e.g. number of molecules of each element)
4. There is a chemical reaction with a chemical equation:
 
$$\text{molecule} + \text{molecule} \rightarrow \text{molecule} + \text{molecule}$$
  - a. The number of atoms of each element in the reaction (e.g. number of protons) of the elements
  - b. All of the reactants and products
  - c. The number of molecules of each element in the reaction (e.g. number of molecules of each element)
  - d. All of the above
5. If the chemical equation is balanced, the number of \_\_\_\_\_ is the same on both sides.
  - a. The number of atoms of each element in the reaction (e.g. number of protons) of the elements
  - b. All of the reactants and products
  - c. The number of molecules of each element in the reaction (e.g. number of molecules of each element)
  - d. The number of molecules of each element in the reaction (e.g. number of molecules of each element)

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**Ralph Thomas Myers**



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