# Nuclear Chemistry: Part 1 Unit Test Review and Test

An article about half-lives describes a daughter isotope. What is a daughter isotope? - ✓ ✓ the isotope that forms from the radioactive decay of a less stable isotope

Examine the nuclear reaction:  $1/1H + 1/0n \rightarrow 2/1H$ . Why is this classified as a nuclear reaction rather than a chemical reaction? -  $\sqrt{\sqrt{A}}$  change has occurred in a nucleus.

Consider the nuclear equation below.

239/94Pu -> X + 4/2He

What is X? - √√235/92U

A sample contains 16.75 g of the radioisotope U-236 and 50.25 g of its daughter isotope, Th-232. How long did it take for decay to take place if one half-life of U-236 is 23 million years? - ✓ ✓ 46 million years

Which represents a balanced nuclear equation? - √√B) 24/11Na -> 24/12Mg + 0/-1e

An isotope undergoes radioactive decay by emitting radiation that has no mass. What other characteristic does the radiation have? -  $\checkmark$  no charge

A radioactive nuclide has 53 protons and 78 neutrons. Which is the symbol of this nuclide? - √√131/53 I

What part of an atom is involved in a chemical reaction? - \(\sqrt{electron}\)

A target is shielded from a radioactive sample by a sheet of metal. What kind of radiation can pass through the sheet of metal and hit the target?  $-\sqrt{\sqrt{gamma}}$  rays only

# **Guide Nuclear Chemistry Answers**

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