

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ followed _____ approaches _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

3. Problem

When aluminum chloride reacts with aluminum hydroxide, aluminum chloride is produced



moles of

$AlCl_3$	$Al(OH)_3$	HCl
2	3	6
$AlCl_3$	$Al(OH)_3$	HCl

2. If you start with 4 moles of $AlCl_3$, how many moles of $Al(OH)_3$ will be produced?

2 moles $AlCl_3$ → 3 moles $Al(OH)_3$ and 6 moles HCl → 4 moles $Al(OH)_3$

Episode 801 Note Taking Guide Key

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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