

NAMING MOLECULAR COMPOUNDS

Name _____

Name the following covalent compounds.

1. CO_2 Carbon dioxide
2. CO Carbon monoxide
3. SO_2 Sulfur dioxide
4. SO_3 Sulfur trioxide
5. N_2O Dinitrogen oxide
6. NO nitrogen oxide
7. N_2O_3 Dinitrogen trioxide
8. NO_2 Nitrogen dioxide
9. N_2O_4 Dinitrogen tetroxide
10. N_2O_5 Dinitrogen Pentoxide
11. PCl_3 Phosphorus trichloride
12. PCl_5 Phosphorus pentachloride
13. NH_3 Nitrogen trihydride
14. SCl_6 Sulfur hexachloride
15. P_2O_5 Diphosphorus pentoxide
16. CCl_4 Carbon tetrachloride
17. SiO_2 Silicon dioxide
18. CS_2 Carbon disulfide
19. OF_2 oxygen difluoride
20. PBr_3 Phosphorus tribromide

- 1- mono
- 2- di
- 3- tri
- 4- tetra
- 5- penta
- 6- hexa

Naming Binary Covalent Compounds Answer Key

Carson-Dellosa Publishing

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Carson-Dellosa Publishing text.

Naming Binary Covalent Compounds Answer Key:

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General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved

exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material *Physical Science* Carson-Dellosa Publishing,2015-03-09 Physical Science for grades 5 to 12 is designed to aid in the review and practice of physical science topics Physical Science covers topics such as scientific measurement force and energy matter atoms and elements magnetism and electricity The book includes realistic diagrams and engaging activities to support practice in all areas of physical science The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series is aligned to current science standards LET REVIEWER NEW CURRICULUM , Prepare with confidence for the Licensure Examination for Teachers LET with this comprehensive and up to date LET Reviewer New Curriculum Specially designed to align with the latest changes in the teacher education curriculum this reviewer is your essential companion in mastering core concepts sharpening test taking skills and boosting your chances of passing the exam Inside you ll find Updated content based on the latest LET framework and curriculum revisions Coverage of General Education Professional Education and Specialization subjects Practice questions with detailed explanations and answer keys Tips and strategies for effective review and exam preparation Suitable for both Elementary and Secondary Level takers Whether you re a first time taker or a repeater aiming to succeed this reviewer is structured to guide you every step of the way Empower yourself with the knowledge and confidence you need to achieve your goal of becoming a licensed professional teacher Start your LET journey the right way with the right reviewer General Organic and Biological Chemistry Kenneth W. Raymond,2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features Student Study Guide to accompany Chemistry Martin Silberberg,Libby Weberg,2005-01-06

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Principles of Modern Chemistry David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom *Survival Guide to General Chemistry* Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many

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<https://staging.conocer.cide.edu/data/scholarship/fetch.php/Ford%20Ranger%20Wiring%20Charging%20System.pdf>

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