

Sous la direction de Jean-Baptiste Legavre

L'informel pour informer Les journalistes et leurs sources



© Nathalie Dumusois

Les journalistes aiment rappeler que leur métier se construit dans la distance. Distance avec les pouvoirs, distance avec les « sources » – celles qui donnent des informations aux gens de presse. Mais comment obtenir des informations qui ne seraient pas le reflet des seuls discours des institutions et de leurs stratégies de communication ? Par un apparent paradoxe, les journalistes n'ont d'autre choix que de quitter, au moins un temps, les habits commodes (et utiles) des défenseurs-de-la-démocratie pour se rapprocher de leurs sources. Tout le paradoxe est là : être « au contact » des sources n'est pas sans risque – le spectre de la « connivence » n'est jamais loin – mais permet aussi de recueillir de précieuses informations pour rendre compte. C'est bien ici que prend place ce qu'il convient d'appeler *l'informel*. Le mot permet d'éclairer un ensemble de pratiques qui s'opposent *a priori* à la dimension officielle, publique, du métier, d'indiquer et de comprendre des écarts, des tensions, des intrications. Il fait entrer qui s'y essaie dans une zone grise occupée par journalistes et informateurs, une zone faite de compromis, d'essais et d'erreurs. Mais il enrichit la compréhension de l'activité, en laissant aux moralistes le rôle (commode) de dénoncer, loin de la bataille.

Jean-Baptiste Legavre est professeur des universités en sciences de l'information et de la communication à l'Université Panthéon-Assas (Paris II). Il dirige l'Ecole de journalisme de l'Institut français de presse.

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver I^+ ions regardless of the source of these ions these molecular and

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the complete ionic equation for this reaction is as follows equation 4 $13 \text{ 2Ag}^+ \text{aq} + 2\text{F}^- \text{aq} + 2\text{NH}_4^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7 \text{s} + 2\text{NH}_4^+ \text{aq} + 2\text{F}^- \text{aq}$ because two $\text{NH}_4^+ \text{aq}$

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