

Syllabus component	Recommended teaching hours	
	SL	HL
<b>Core</b> 1. Stoichiometric relationships 2. Atomic structure 3. Periodicity 4. Chemical bonding and structure 5. Energetics/thermochemistry 6. Chemical kinetics 7. Equilibrium 8. Acids and bases 9. Redox processes 10. Organic chemistry 11. Measurement and data processing	<b>95</b> 13.5 6 6 13.5 9 7 4.5 6.5 8 11 10	
<b>Additional higher level (AHL)</b> 12. Atomic structure 13. The periodic table—the transition metals 14. Chemical bonding and structure 15. Energetics/thermochemistry 16. Chemical kinetics 17. Equilibrium 18. Acids and bases 19. Redox processes 20. Organic chemistry 21. Measurement and analysis		<b>60</b> 2 4 7 7 6 4 10 6 12 2
<b>Option</b> A. Materials B. Biochemistry C. Energy D. Medicinal chemistry	<b>15</b> 15 15 15	<b>25</b> 25 25 25

# Ib Chemistry Sl Mark Scheme 2013

**Daniel F McAuley**



## **Ib Chemistry Sl Mark Scheme 2013:**

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