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SL Paper 2

A student determined the percentage of the active ingredient magnesium hydroxide, $\text{Mg}(\text{OH})_2$, in a 1.24 g antacid tablet.

The antacid tablet was added to 50.00 cm^3 of $0.100 \text{ mol dm}^{-3}$ sulfuric acid, which was in excess.

- Calculate the amount, in mol, of H_2SO_4 . [1]
- Formulate the equation for the reaction of H_2SO_4 with $\text{Mg}(\text{OH})_2$. [1]
- The excess sulfuric acid required 20.60 cm^3 of $0.1133 \text{ mol dm}^{-3}$ NaOH for neutralization.
Calculate the amount of excess acid present. [1]
- Calculate the amount of H_2SO_4 that reacted with $\text{Mg}(\text{OH})_2$. [1]
- Determine the mass of $\text{Mg}(\text{OH})_2$ in the antacid tablet. [2]
- Calculate the percentage by mass of magnesium hydroxide in the 1.24 g antacid tablet to three significant figures. [1]

Phosphoric acid, H_3PO_4 , can undergo stepwise neutralization, forming amphiprotic species.

- Formulate an equation for the reaction of one mole of phosphoric acid with one mole of sodium hydroxide. [1]
- Formulate **two** equations to show the amphiprotic nature of H_2PO_4^- . [2]
- Calculate the concentration of H_3PO_4 if 25.00 cm^3 is completely neutralised by the addition of 28.40 cm^3 of $0.5000 \text{ mol dm}^{-3}$ NaOH . [2]
- Outline the reason that sodium hydroxide is considered a Brønsted–Lowry base. [1]

Iron may be extracted from iron (II) sulfide, FeS .

Iron (II) sulfide, FeS , is ionically bonded.

The first step in the extraction of iron from iron (II) sulfide is to roast it in air to form iron (III) oxide and sulfur dioxide.

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Thirumalaisamy P.
Velavancorresponding



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Ib Chemistry 2013 Sl Past Paper Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be more evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **Ib Chemistry 2013 Sl Past Paper** , a literary masterpiece that delves deep to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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