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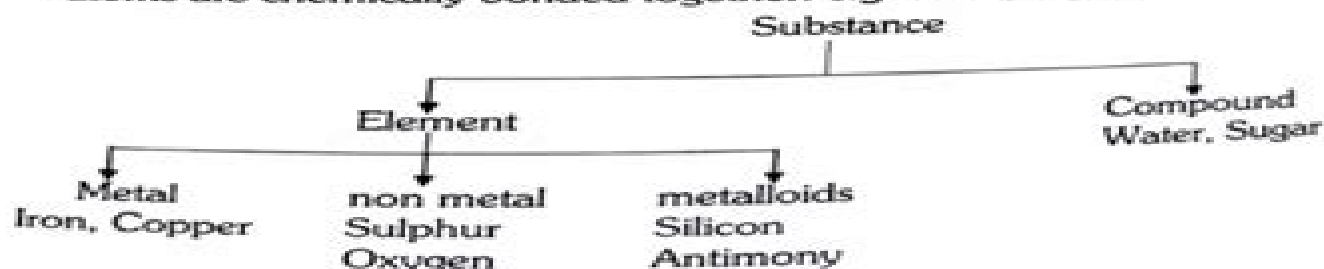
Chapter- Metals and non Metals,

Some Basic Concepts

1. Pure substance- Pure substance can either be made of same type of atoms or they may be made up of two or more types of atoms.

Pure substance are of two types

- (a) Elements- A substance that is made up of only one kind of atoms and cannot be broken further into smaller components is called an element.
Element is further classified as metal, non metals and metalloids. e.g Iron, Copper.
- (b) Compounds- A compound is a substance formed when 2 or more different atoms are chemically bonded together. e.g Water, sugar.



(Class - VIII : SCIENCE)

Common elements and their symbols

English name of the element		Symbol	English name of the element		Symbol
1.	Argon	Ar	15.	Iodine	I
2.	Arsenic	As	16.	Lithium	Li
3.	Aluminium	Al	17.	Magnesium	Mg
4.	Boron	B	18.	Manganese	Mn
5.	Barium	Ba	19.	Nitrogen	N
6.	Bromine	Br	20.	Neon	Ne
7.	Carbon	C	21.	Nickel	Ni
8.	Calcium	Ca	22.	Oxygen	O
9.	Chlorine	Cl	23.	Phosphorus	P
10.	Chromium	Cr	24.	Platinum	Pt
11.	Cobalt	Co	25.	Sulphur	S
12.	Fluorine	F	26.	Uranium	U
13.	Hydrogen	H	27.	Zinc	Zn
14.	Helium	He			

TABLE 2. Symbols derived from Latin names

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Table of Contents Metals Section Ch 19 Answers

1. Understanding the eBook Metals Section Ch 19 Answers
 - The Rise of Digital Reading Metals Section Ch 19 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Metals Section Ch 19 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Metals Section Ch 19 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Metals Section Ch 19 Answers
 - Personalized Recommendations
 - Metals Section Ch 19 Answers User Reviews and Ratings
 - Metals Section Ch 19 Answers and Bestseller Lists
5. Accessing Metals Section Ch 19 Answers Free and Paid eBooks

- Metals Section Ch 19 Answers Public Domain eBooks
 - Metals Section Ch 19 Answers eBook Subscription Services
 - Metals Section Ch 19 Answers Budget-Friendly Options
6. Navigating Metals Section Ch 19 Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Metals Section Ch 19 Answers Compatibility with Devices
 - Metals Section Ch 19 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Metals Section Ch 19 Answers
 - Highlighting and Note-Taking Metals Section Ch 19 Answers
 - Interactive Elements Metals Section Ch 19 Answers
8. Staying Engaged with Metals Section Ch 19 Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Metals Section Ch 19 Answers
9. Balancing eBooks and Physical Books Metals Section Ch 19 Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Metals Section Ch 19 Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Metals Section Ch 19 Answers
- Setting Reading Goals Metals Section Ch 19 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Metals Section Ch 19 Answers
- Fact-Checking eBook Content of Metals Section Ch 19 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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