# Hydrocarbons

Methane	General formula for an alkane		General formula for an alkene	Ethene	Propene	Butene	
	How to test between an alkene and alkane		Alkane or alkene?  C/H <sub>4</sub> C/H <sub>6</sub> C/H <sub>6</sub> C/H <sub>6</sub> C/H <sub>6</sub>				
Ethane				Show how ethere is polymerised			
	Definition of a hydrocarbon						
Propune		What is meant by saturated/unsaturated		Draw the monomer that makes this polymer  Br H Br H Br H  I I I I I  H-C-C-C-C-C-C-H  I I I I I  H H H H H H			
Butane		How ethanol is produced from ethane		Describe the proce Process:	ns of cracking	Complete the equations for cracking  C <sub>0</sub> H <sub>10</sub> → C <sub>1</sub> H <sub>4</sub> + + C <sub>4</sub> H <sub>38</sub>	
A way to remember the order of alkanes		_by fermentation		Types of products.  Conditions:  Why it is done:	The types of products produced:		

# **Hydrocarbon Compounds Core Teaching Resources**

**TD Snyder** 

**Hydrocarbon Compounds Core Teaching Resources:** 

Uncover the mysteries within is enigmatic creation, Embark on a Mystery with **Hydrocarbon Compounds Core Teaching Resources**. This downloadable ebook, shrouded in suspense, is available in a PDF format (\*). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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