

Equation:

$$f(t) = a\left(\frac{1}{2}\right)^{\frac{t}{h}}$$

a = initial amount

t = time

h = half-life time

$f(t)$ = remains of material

The half-life of Po-214 is 0.001 seconds.

How much of a 10g sample is left after 0.003 seconds?

A: 10g

t: 0.003 secs

h: 0.001 secs

$$f(0.003) = 10\left(\frac{1}{2}\right)^{\frac{0.003}{0.001}}$$

$$f(0.003) = 1.25$$

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