



# Haier Dw100s Tu Dishwashers Owners Manual

**Graham Dixon**



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$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web the net ionic equation for equation 1 is given as equation 3  $\text{s}^{2-} \text{ aq}$   $\text{Fe}^{2+} \text{ aq}$   $\text{gt}$   $\text{FeS} \text{ s}$  eq 3 8 note that the phase indicator aq is optional on ionic species but is

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web  $\text{CuClO}_4$  is soluble all perchlorates are soluble while  $\text{PbSO}_4$  is insoluble sulfates are generally soluble except for  $\text{Pb}^{2+}$   $\text{Ba}^{2+}$   $\text{Hg}_2^{2+}$  and  $\text{Sr}^{2+}$  the net ionic equation is

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