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Guillaume Favre



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following equation $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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$\text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_2 \rightarrow 2 \text{ZnCl}_2 + 2 \text{Cu}$ how many moles of $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_2$ will be produced from 23.0 g of $\text{Ce} + \text{Zn}$

assuming $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_2$ is available in excess

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$\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$ br

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