Far-Eastern University
Institute of Arts and Sciences
Department of Medical Technology
Nicanor Reyes St., Sampaloc, Manila

MT 111 RED-ANALYTICAL CHEMISTRY

Name: _____ Section: _____ Score: _____

CHAPTER 8 WORKSHEET

1. Balance the following reactions in both acidic and basic media.

   a. $MnO_4^{-} + H_2O_2 \rightarrow MnO_2 + O_2^{-}$

   To balance this, we should separate this into two half-equations which are $MnO_4^{-} \rightarrow MnO_2$ and $H_2O_2 \rightarrow O_2^{-}$ which are balanced below.

   Balancing the reduction reaction

$$MnO_4^{-} \rightarrow MnO_2$$
$$MnO_4^{-} \rightarrow 2\ MnO_2$$
$$2\ H^{+} + MnO_4^{-} \rightarrow 2\ MnO_2$$

   Balancing the oxidation reaction

$$H_2O_2 \rightarrow O_2^{-}$$
$$H_2O_2 \rightarrow 2\ O_2^{-}$$
$$H_2O_2 \rightarrow 2\ O_2^{-} + 2\ H^{+}$$

   In combining the two half-equations, make sure that the electrons are equal in each half-equation.

$$2\ H^{+} + MnO_4^{-} \rightarrow 2\ MnO_2$$
$$H_2O_2 \rightarrow 2\ O_2^{-} + 2\ H^{+}$$
$$2\ H^{+} + H_2O_2 + MnO_4^{-} \rightarrow 2\ MnO_2 + 2\ O_2^{-} + 2\ H^{+}$$
$$H_2O_2 + MnO_4^{-} \rightarrow 2\ MnO_2 + 2\ O_2^{-} \qquad \text{(acidic)}$$

   To make this in basic medium, we add enough OH^{-} equal to H^{+}.

$$2\ OH^{-} + H_2O_2 + MnO_4^{-} \rightarrow 2\ MnO_2 + 2\ O_2^{-} + 2\ OH^{-}$$
$$2\ OH^{-} + H_2O_2 + MnO_4^{-} \rightarrow 2\ MnO_2 + 2\ H_2O \qquad \text{(basic)}$$

   b. $Cr_2O_7^{2-} + I^{-} \rightarrow Cr^{3+} + I_2$

   To balance this, we should separate this into two half-equations which are $Cr_2O_7^{2-} \rightarrow Cr^{3+}$ and $I^{-} \rightarrow I_2$ which are balanced below.

   Balancing the reduction reaction

$$Cr_2O_7^{2-} \rightarrow Cr^{3+}$$
$$Cr_2O_7^{2-} \rightarrow 2\ Cr^{3+}$$
$$Cr_2O_7^{2-} \rightarrow 2\ Cr^{3+} + 7\ H_2O$$
$$14\ H^{+} + Cr_2O_7^{2-} \rightarrow 2\ Cr^{3+} + 7\ H_2O$$
$$6\ e^{-} + 14\ H^{+} + Cr_2O_7^{2-} \rightarrow 2\ Cr^{3+} + 7\ H_2O$$

   Balancing the oxidation reaction

$$I^{-} \rightarrow I_2$$

Episode 1103 Chemistry Worksheet

Daniela Niemeyer

Episode 1103 Chemistry Worksheet:

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