

Question 3 : Extended Questions

(16 Marks)

(1)

32 grams of sulfur combine with 32 grams of oxygen to form sulfur oxide.

(2marks)

a. Calculate the empirical formula. (A_r : S=32 , O=16)

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b. Define the Avogadro's constant.

(1mark)

.....

.....

c. How many moles of molecules are there in 54 grams of water?

(2 marks)

.....

.....

.....

.....

d. What is the volume of 1.7 gram of ammonia at rtp?

(1 mark)

(A_r : N=14 / H=1)

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(2)

(6marks)

a. Define strong acid with example.

(1marks)

.....

.....

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Table of Contents Mpumalanga Final Exam Grade 9 2013

1. Understanding the eBook Mpumalanga Final Exam Grade 9 2013
 - The Rise of Digital Reading Mpumalanga Final Exam Grade 9 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Mpumalanga Final Exam Grade 9 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mpumalanga Final Exam Grade 9 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mpumalanga Final Exam Grade 9 2013
 - Personalized Recommendations
 - Mpumalanga Final Exam Grade 9 2013 User Reviews and Ratings
 - Mpumalanga Final Exam Grade 9 2013 and Bestseller Lists
5. Accessing Mpumalanga Final Exam Grade 9 2013 Free and Paid eBooks
 - Mpumalanga Final Exam Grade 9 2013 Public Domain eBooks
 - Mpumalanga Final Exam Grade 9 2013 eBook Subscription Services
 - Mpumalanga Final Exam Grade 9 2013 Budget-Friendly Options

6. Navigating Mpumalanga Final Exam Grade 9 2013 eBook Formats
 - ePub, PDF, MOBI, and More
 - Mpumalanga Final Exam Grade 9 2013 Compatibility with Devices
 - Mpumalanga Final Exam Grade 9 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mpumalanga Final Exam Grade 9 2013
 - Highlighting and Note-Taking Mpumalanga Final Exam Grade 9 2013
 - Interactive Elements Mpumalanga Final Exam Grade 9 2013
8. Staying Engaged with Mpumalanga Final Exam Grade 9 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mpumalanga Final Exam Grade 9 2013
9. Balancing eBooks and Physical Books Mpumalanga Final Exam Grade 9 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mpumalanga Final Exam Grade 9 2013
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mpumalanga Final Exam Grade 9 2013
 - Setting Reading Goals Mpumalanga Final Exam Grade 9 2013
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mpumalanga Final Exam Grade 9 2013
 - Fact-Checking eBook Content of Mpumalanga Final Exam Grade 9 2013
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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