

1. सभी प्रश्नों के उत्तर दीजिए :

1.(a) प्रारंभिक पंक्ति संक्रिया का इस्तेमाल करते हुए, आव्यूह

$$A = \begin{bmatrix} 1 & 3 & 1 \\ 2 & -1 & 7 \\ 3 & 2 & -1 \end{bmatrix}$$

का प्रतिलोम मालूम कीजिए। अतएव रैखिक समीकरणों के तंत्र को हल कीजिए

$$x + 3y + z = 10$$

$$2x - y + 7z = 21$$

$$3x + 2y - z = 4$$

1.(b) लीजिए एक वर्ग आव्यूह A और उसका संलग्न A^* । दर्शाइए कि आव्यूहों AA^* और A^*A के आइगन मान वास्तविक हैं। आगे दर्शाइए कि $\text{ट्रेस}(AA^*) = \text{ट्रेस}(A^*A)$ ।

1.(c) मूल्यांकन कीजिए $\int_0^1 \left(2x \sin \frac{1}{x} - \cos \frac{1}{x} \right) dx$ ।

1.(d) एक ऐसे समतल का समीकरण मालूम कीजिए जो बिंदुओं $(0, 1, 1)$ और $(2, 0, -1)$ में से गुजरता हो और जो बिंदुओं $(-1, 1, -2)$, $(3, -2, 4)$ को जोड़ने वाली रेखा के समांतर हो। इसके साथ ही रेखा और समतल के बीच की दूरी भी मालूम कीजिए।

1.(e) एक गोलक S के व्यास के आमने-सामने के सिरों पर बिंदु $(0, 1, 0)$, $(3, -5, 2)$ हैं। गोलक के समीकरण को मालूम कीजिए, जिसका गोलक S का समतल $5x - 2y + 4z + 7 = 0$ के साथ प्रतिच्छेद एक वृहत् वृत्त के रूप में है।

2.(a)(i) लीजिए कि P_n अधिकांश n पर कोटि के सभी वास्तविक बहुपदों की सदिश समष्टि को छोड़ित करता है और कि $T: P_2 \rightarrow P_3$ निम्नलिखित द्वारा दत्त एक रैखिक रूपांतरण है :

$$T(p(x)) = \int_0^x p(t) dt, \quad p(x) \in P_2$$

P_2 और P_3 के क्रमशः $\{1, x, x^2\}$ और $\{1, x, 1+x^2, 1+x^3\}$ आधारों के विषय में T का आव्यूह मालूम कीजिए। इस के साथ T की शून्य समष्टि भी मालूम कीजिए।

2.(a)(ii) लीजिए कि V एक n -विमीय सदिश समष्टि है और $T: V \rightarrow V$ एक व्युत्क्रमणीय रैखिक संकारक है। यदि $\beta = \{X_1, X_2, \dots, X_n\}$ आधार हो V का, तो दर्शाइए कि $\beta' = \{TX_1, TX_2, \dots, TX_n\}$ भी V का आधार होगा।

2.(b)(i) लीजिए कि $A = \begin{bmatrix} 1 & 1 & 1 \\ 1 & \omega^2 & \omega \\ 1 & \omega & \omega^2 \end{bmatrix}$ जहाँ $\omega (\neq 1)$ एक का घनमूल है। यदि $\lambda_1, \lambda_2, \lambda_3$ चयनित करते हैं A^2 के आइगन मानों का, तो दर्शाइए कि $|\lambda_1| + |\lambda_2| + |\lambda_3| \leq 9$ ।

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made by antoine laurent lavoisier at the end of the 18th century lavoisier supposed that all acids must contain oxygen and this idea was incorporated in the names used for this element in the various languages the english

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web work done by svante arrhenius in the late 19th century set the groundwork for our current understanding of acid base theory despite the relatively limited technology available to him he was able to develop theories that adequately explained the properties and chemical reactivity of many acids and bases

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web the A^- ion is a base because it is accepting a proton from the H_2O the reversible reaction contains two acids and two

bases we think of them in pairs called conjugate pairs when the acid ha loses a proton it forms a base a when the base a accepts a proton back again it obviously refoms the acid ha

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