

Atoms and Bonding → Review and Reinforce

Ionic Bonds**Understanding Main Ideas**

Answer the following questions on a separate sheet of paper.

- How does an atom become a positive ion? How does an atom become a negative ion?
- How do ions form electrically neutral compounds?
- What characteristics do ionic compounds show?
- Why does the electrical conductivity of ionic compounds change when they are dissolved in water?

Ions and Their Charges		
Name	Charge	Symbol/Formula
Ammonium	1+	NH_4^+
Potassium	1+	K^+
Calcium	2+	Ca^{2+}
Magnesium	2+	Mg^{2+}
Chloride	1-	Cl^-
Oxide	2-	O^{2-}
Sulfide	2-	S^{2-}
Phosphate	3-	PO_4^{3-}

Use the chart above to answer the following on a separate sheet of paper.

- How many potassium ions are needed to balance the charge of one sulfide ion? Explain.
- Predict the formulas for calcium chloride and potassium phosphate.
- Name the following compounds: MgS , NH_4Cl , and K_2O .
- Which ions in the table are polyatomic ions?

Building Vocabulary

Answer the following questions on a separate sheet of paper.

- What is an ion?
- What is an ionic bond?
- What is the arrangement of ions in a crystal?

Ionic Bonds Review And Reinforce Section 4 1

S Ben Porath



Ionic Bonds Review And Reinforce Section 4 1:

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