

Atoms and Bonding → Review and Reinforce

Ionic Bonds**Understanding Main Ideas**

Answer the following questions on a separate sheet of paper.

- How does an atom become a positive ion? How does an atom become a negative ion?
- How do ions form electrically neutral compounds?
- What characteristics do ionic compounds show?
- Why does the electrical conductivity of ionic compounds change when they are dissolved in water?

Ions and Their Charges		
Name	Charge	Symbol/Formula
Ammonium	1+	NH_4^+
Potassium	1+	K^+
Calcium	2+	Ca^{2+}
Magnesium	2+	Mg^{2+}
Chloride	1-	Cl^-
Oxide	2-	O^{2-}
Sulfide	2-	S^{2-}
Phosphate	3-	PO_4^{3-}

Use the chart above to answer the following on a separate sheet of paper.

- How many potassium ions are needed to balance the charge of one sulfide ion? Explain.
- Predict the formulas for calcium chloride and potassium phosphate.
- Name the following compounds: MgS , NH_4Cl , and K_2O .
- Which ions in the table are polyatomic ions?

Building Vocabulary

Answer the following questions on a separate sheet of paper.

- What is an ion?
- What is an ionic bond?
- What is the arrangement of ions in a crystal?

Ionic Bond Review And Reinforce

R Sanford



Ionic Bond Review And Reinforce:

The Enigmatic Realm of **Ionic Bond Review And Reinforce**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Ionic Bond Review And Reinforce** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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