

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	CHARGE OF ION (POSITIVE OR NEGATIVE)	TYPE OF ION
1	Sodium	Na^+	11	10	positive	cation
2	Chlorine	Cl^-	17	18	negative	anion
3	Strontium	Sr^{2+}	38	36	positive	cation
4	Phosphorus	P^{3-}	15	18	negative	anion
5	Calcium	Ca^{2+}	20	18	positive	cation
6	Barium	Ba^{2+}	56	54	positive	cation
7	Fluorine	F^-	9	10	negative	anion
8	Lithium	Li^+	3	2	positive	cation
9	Magnesium	Mg^{2+}	12	10	positive	cation
10	Sulfur	S^{2-}	16	18	negative	anion
11	Aluminum	Al^{3+}	13	10	positive	cation
12	Oxygen	O^{2-}	8	10	negative	anion
13	Boron	B^{3+}	5	2	positive	cation
14	Strontium	Sr^{2+}	38	36	positive	cation
15	Calcium	Ca^{2+}	20	18	positive	cation

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Table of Contents Ion Practice Set Og Answers

1. Understanding the eBook Ion Practice Set Og Answers
 - The Rise of Digital Reading Ion Practice Set Og Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ion Practice Set Og Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ion Practice Set Og Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ion Practice Set Og Answers
 - Personalized Recommendations
 - Ion Practice Set Og Answers User Reviews and Ratings
 - Ion Practice Set Og Answers and Bestseller Lists
5. Accessing Ion Practice Set Og Answers Free and Paid eBooks
 - Ion Practice Set Og Answers Public Domain eBooks
 - Ion Practice Set Og Answers eBook Subscription Services
 - Ion Practice Set Og Answers Budget-Friendly Options
6. Navigating Ion Practice Set Og Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ion Practice Set Og Answers Compatibility with Devices
 - Ion Practice Set Og Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ion Practice Set Og Answers
 - Highlighting and Note-Taking Ion Practice Set Og Answers
 - Interactive Elements Ion Practice Set Og Answers
8. Staying Engaged with Ion Practice Set Og Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Ion Practice Set Og Answers
- 9. Balancing eBooks and Physical Books Ion Practice Set Og Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ion Practice Set Og Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ion Practice Set Og Answers
 - Setting Reading Goals Ion Practice Set Og Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ion Practice Set Og Answers
 - Fact-Checking eBook Content of Ion Practice Set Og Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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