

Chemistry Periodic Table Test Chapter 6 Chem IH

Matching Match each item with the correct statement below.

- | | |
|----------------------|---------------------|
| a. electronegativity | e. transition metal |
| b. ionization energy | f. periodic law |
| c. electrons | g. group |
| d. metal | h. period |

- ____ 1. vertical column in the periodic table
- ____ 2. type of element characterized by the presence of electrons in the *d* orbital
- ____ 3. type of element characterized by losing electrons easily
- ____ 4. ability of an atom to attract electrons when the atom is in a compound
- ____ 5. energy required to remove an electron from an atom

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- ____ 6. What is another name for the main group elements?
- | | |
|------------------------------|-------------------------|
| a. s block elements | c. s & p block elements |
| b. p block elements elements | d. d block elements |
- ____ 7. What is another name for the transition metals?
- | | |
|------------------------------|------------------------------|
| a. noble gases | c. p block elements elements |
| b. s block elements elements | d. d block elements |
- ____ 8. Who arranged the elements according to atomic mass and used the arrangement to predict the properties of missing elements?
- | | |
|----------------------|---------------------|
| a. Henry Moseley | c. John Dalton |
| b. Antoine Lavoisier | d. Dmitri Mendeleev |
- ____ 9. Which of the following categories includes the majority of the elements?
- | | |
|---------------|--------------|
| a. metalloids | c. metals |
| b. liquids | d. nonmetals |
- ____ 10. In which of the following sets is the symbol of the element, the number of protons, and the number of electrons given correctly?
- | | |
|---------------------------------|------------------------------------|
| a. In, 49 protons, 49 electrons | c. Cs, 55 protons, 132.9 electrons |
| b. Zn, 30 protons, 60 electrons | d. F, 19 protons, 19 electrons |
- ____ 11. The atomic number of an element is the total number of which particles in the nucleus?
- | | |
|-------------|--------------------------|
| a. neutrons | c. electrons |
| b. protons | d. protons and electrons |
- ____ 12. What element has the electron configuration $1s^2 2s^2 2p^6 3s^2 3p^2$?
- | | |
|-------------|------------|
| a. nitrogen | c. silicon |
| b. selenium | d. silver |
- ____ 13. Which subatomic particle plays the greatest part in determining the properties of an element?
- | | |
|-------------|----------------------|
| a. proton | c. neutron |
| b. electron | d. none of the above |
- ____ 14. Which of the following elements is a transition metal?
- | | |
|-----------|--------------|
| a. cesium | c. tellurium |
| b. copper | d. tin |

Holt Chemistry The Periodic Table Chapter Test

David Kirk



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