

Q2) Attempt any four of the following:

[16]

- a) Describe the nature of liquid according to kinetic molecular model.
- b) Explain physical or Vander Waal's adsorption.
- c) Write the equation of the line passing through a point
 - i) $(-4, 0)$ and having slope - 3. &
 - ii) $(-2, 0)$ & slope - $\frac{1}{3}$.
- d) Explain de-Broglies hypothesis and derive the expression for wavelength in terms of kinetic energy of particle.
- e) Explain the importance of Carnot's cycle in terms of
 - i) determination of efficiency.
 - ii) thermodynamic scale of temperature.
- f) Give assumptions and limitations of Bohr's theory.

Q3) Answer any four of the following:

[16]

- a) i) If $Y = \frac{1-x^2}{1+x^2}$, find $\frac{dy}{dx}$.
- ii) Solve the integral $\int (2x)^{\frac{3}{2}} \cdot dx$

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Orin Fletcher Stafford



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Report on the examination in chemistry, 1993 **Chemistry** Ying-wai Chan, 1981 **Chemistry Syllabus** Pearson Custom Publishing, 1994-05-01 *Practical Chemistry* J. Howard, 1873 **Chemistry syllabus** Caribbean Examination Council, 1987 **Chemistry for Degree Students B.Sc. Semester - IV (As per CBCS)** Madan R.L., 2017 This textbook has been designed to meet the needs of B Sc Fourth Semester students of Chemistry as per the UGC Choice Based Credit System CBCS With its traditional approach to the subject this textbook lucidly explains principles of chemistry Important topics such as transition elements coordination chemistry crystal field theory kinetic theory of gases liquids solids and chemical kinetics are aptly discussed to give an overview of inorganic and physical chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures

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1. Understanding the eBook Fybsc Chemistry Syllabus 2015
 - The Rise of Digital Reading Fybsc Chemistry Syllabus 2015
 - Advantages of eBooks Over Traditional Books
2. Identifying Fybsc Chemistry Syllabus 2015
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Fybsc Chemistry Syllabus 2015
 - User-Friendly Interface
4. Exploring eBook Recommendations from Fybsc Chemistry Syllabus 2015
 - Personalized Recommendations
 - Fybsc Chemistry Syllabus 2015 User Reviews and Ratings
 - Fybsc Chemistry Syllabus 2015 and Bestseller Lists

5. Accessing Fybsc Chemistry Syllabus 2015 Free and Paid eBooks
 - Fybsc Chemistry Syllabus 2015 Public Domain eBooks
 - Fybsc Chemistry Syllabus 2015 eBook Subscription Services
 - Fybsc Chemistry Syllabus 2015 Budget-Friendly Options
6. Navigating Fybsc Chemistry Syllabus 2015 eBook Formats
 - ePub, PDF, MOBI, and More
 - Fybsc Chemistry Syllabus 2015 Compatibility with Devices
 - Fybsc Chemistry Syllabus 2015 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Fybsc Chemistry Syllabus 2015
 - Highlighting and Note-Taking Fybsc Chemistry Syllabus 2015
 - Interactive Elements Fybsc Chemistry Syllabus 2015
8. Staying Engaged with Fybsc Chemistry Syllabus 2015
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Fybsc Chemistry Syllabus 2015
9. Balancing eBooks and Physical Books Fybsc Chemistry Syllabus 2015
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Fybsc Chemistry Syllabus 2015
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Fybsc Chemistry Syllabus 2015
 - Setting Reading Goals Fybsc Chemistry Syllabus 2015
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Fybsc Chemistry Syllabus 2015
 - Fact-Checking eBook Content of Fybsc Chemistry Syllabus 2015
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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