

CHEMISTRY

2013

Time: 30 minutes

Max. Marks: 15

SECTION "A" (MULTIPLE CHOICE QUESTIONS)

1. Choose the correct answer for each from the given options:

- (i) it is a trial idea:
(a) Experiment (b) Hypothesis (c) Theory (d) Scientific Law
- (ii) It is the empirical formula of glucose:
(a) CH₂O (b) C₂H₄O₂ (c) C₅H₁₀O₅ (d) C₆H₁₂O₆
- (iii) It is not an isotope of oxygen:
(a) ¹⁶₈O (b) ¹⁷₈O (c) ¹⁸₈O (d) ¹⁹₈O
- (iv) The vertical columns of the periodic table are called:
(a) Periods (b) Series (c) Groups (d) Rows
- (v) It is an ionic compound:
(a) H₂O (b) CO₂ (c) NH₃ (d) NaCl
- (vi) Neither definite shape nor definite volume is the property of:
(a) Solid (b) Liquid (c) Gas (d) None of them
- (vii) It is defined as the number of moles of solution dissolved in one litre of solution:
(a) Molarity (b) Molality (c) Normality (d) Mole fraction
- (viii) It is the unit of electrical current:
(a) volt (b) coulomb (c) faraday (d) ampere
- (ix) It is a proton-donor:
(a) Acid (b) Base (c) Salt (d) Alkali
- (x) It is an exothermic process:
(a) Evaporation (b) Respiration
(c) Thermal decomposition (d) Melting
- (xi) It is an ion-exchanger resin:
(a) Na-zeolite (b) Ca(OH)₂ (c) MgSO₄ (d) NaCl
- (xii) When carbon burns in air, this gas is produced:
(a) SO₂ (b) H₂S (c) NH₃ (d) CO₂
- (xiii) It is the study of Hydrocarbons and their derivatives:
(a) Physical Chemistry (b) Inorganic Chemistry
(c) Organic Chemistry (d) Environmental Chemistry
- (xiv) This gas is used for respiration:
(a) Hydrogen (b) Oxygen (c) Nitrogen (d) Chlorine
- (xv) It is the only liquid metal:
(a) Bromine (b) Mercury (c) Gold (d) Sulphur
- (xvi) The most abundant and useful halogen is:
(a) Fluorine (b) Chlorine (c) Bromine (d) Iodine
- (xvii) Red ink is normally prepared from:

May 2013 Chemistry F32past Paper

Thomas Griffiths



May 2013 Chemistry F32past Paper:

Unveiling the Magic of Words: A Report on "**May 2013 Chemistry F32past Paper**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**May 2013 Chemistry F32past Paper**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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