71. Assume that a drop of liquid evaporates by decrease in its surface energy, so that its temperature remains unchanged. What should be the minimum radius of the drop for this to be possible? The surface tension is T, density of liquid is p and L is its latent heat of vaporization.

(1) JT/pL

(2) T/p/L

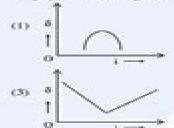
(3) 2T/p/L.

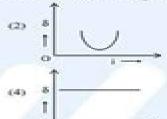
(4) pL/T

Set. (3) $\rho 4\pi R^{2}\Delta RL = T 4\pi [R^{2} - (R - \Delta R)^{2}]$ $\rho R^{2}\Delta RL = T[R^{2} - R^{2} + 2R\Delta R - \Delta R^{2}]$ $\rho R^{2}\Delta RL = T2R\Delta R (\Delta R \text{ is very small})$ $\rho = 2T$



The graph between angle of deviation (δ) and angle of incidence (i) for a triangular prism is represented by:





- Sed. (2)
- Let [6] denote the dimensional formula of the permittivity of vacuum. If M = mass, L = length, T = time and A = electric current, then:

(1)
$$|c_0| = |M^2L^2T^4A^2|$$

(3)
$$[c_n] = [M^{\dagger}L^2T^{\dagger}A]$$

(2)
$$[c_0] = [M^{-1}L^2T^{-1}A^{-2}]$$

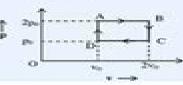
$$(4) [c_0] = [M^3L^3T^2A]$$

O

- Sol. (1)
- 73. $\frac{1}{4\pi\epsilon_0}\frac{q^2}{r^2} = F$

$$c_0 = \frac{[A^2T^2]}{[MLT^{-2}L^2]} = [M^{-1}L^{-3}A^2T^4]$$

74



The above p-v diagram represents the thermodynamic cycle of an engine, operating with an ideal monoatomic gas. The amount of heat extracted from the source in a single cycle is

(1)
$$\left(\frac{13}{2}\right)p_0v_0$$

(2) $\left(\frac{11}{2}\right)p_0v_0$

 $(3).4p_0v_0$

 $(4)\ p_n v_n$

Sed. (2)

Heat is extracted from the source in path DA and AB is

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Yan Bai

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