# Finding the empirical formula

Example: What is the empirical formula of H<sub>2</sub>O<sub>2</sub>?

Answer: Divide everything by the smallest number in the formula – in this case 2. Dividing through by 2 gives HO.

# TASK: Find the empirical formulas of the following:

- 1. C<sub>6</sub>H<sub>6</sub> CH
- C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> CH<sub>2</sub>O
- 3. C<sub>4</sub>H<sub>8</sub> CH<sub>2</sub>
- 4. H<sub>2</sub>O H<sub>2</sub>O
- 5. H<sub>4</sub>C<sub>4</sub>O<sub>8</sub> H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>
- 6. P<sub>4</sub>O<sub>10</sub> P<sub>2</sub>O<sub>5</sub>

- 7. Ca(OH)<sub>2</sub> CaO<sub>2</sub>H<sub>2</sub>
- 8. (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> N<sub>2</sub>H<sub>8</sub>CO<sub>3</sub>
- 9.  $Mg(NO_3)_2$   $MgN_2O_6$

# **How To Calculate Emperical And Molecular Fomula S**

**Brendan G. Carr** 

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