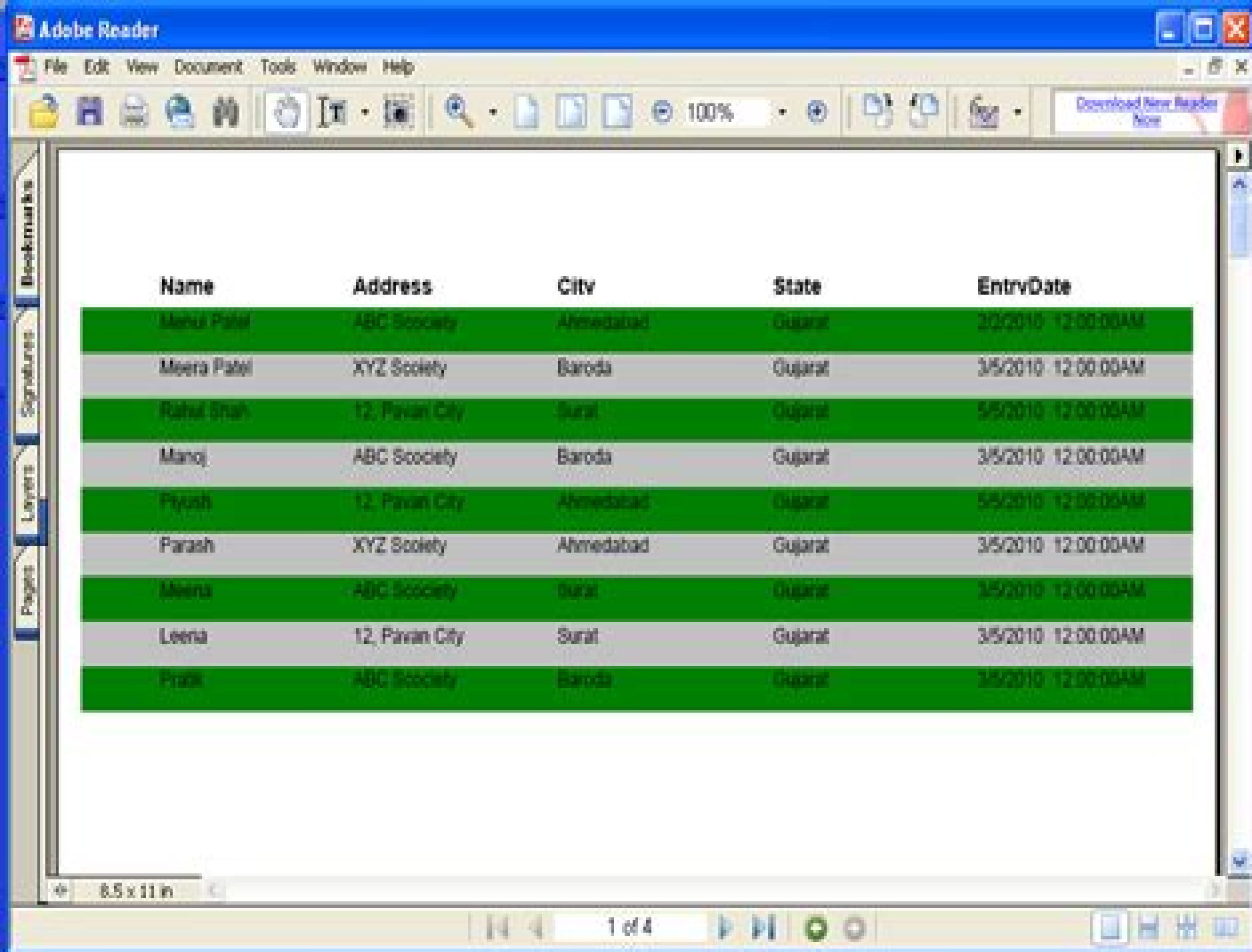




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J Ma



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with the exhibition you will need to provide the total valuation for the show and determine insurance coverage as part of the budgeting process please use the budget template provided here 3

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then $ph = -\log 0.100 = 1.000$ ph $-\log 0.100 = 1.000$ b titrant volume 12.50 ml

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Volume of NaOH added (mL)	pH
4.6	4.4
4.6	5.9
5.9	5.9

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web background plot of the ph of a solution against the volume of titrant added is called a titration curve the ph can be measured directly with a ph meter while titrant is added from a buret for the acids used in this experiment the titrant will always be a 0.1 M solution of the strong base NaOH

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