

Name :

Date :

Periodic Table Trends Worksheet

Use the periodic table and your knowledge of properties and periodic trends to answer the following questions.

1. Ionization Energy

- Define ionization energy.
- What is the trend as you move down a group (column)?
- What is the trend as you move across a period (row)?
- Which family of elements has the highest ionization energies?
- Which family of elements has the lowest ionization energies?
- Arrange the following atoms in order of increasing ionization energy: lithium, oxygen, magnesium, strontium, and chlorine.

2. Electronegativity

- Define electronegativity.
- What is the trend as you move down a group (column)?
- What is the trend as you move across a period (row)?
- Which family of elements has the highest electronegativities?
- Arrange the following atoms in order of increasing ionization energy: lithium, oxygen, magnesium, iodine, and boron.

Honors Chemistry Periodic Trends

RD Boyd



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