

Concept Review

Section: What Is Matter?

1. Classify the following as a homogeneous, *O*, or a heterogeneous, *E*, mixture.

_____ a. a pail of sand and water

_____ b. air

_____ c. human blood

_____ d. banana split

_____ e. chocolate syrup

_____ f. sea water

2. Compare and contrast atoms and molecules.

3. Explain the difference between a pure substance and a homogeneous mixture.

4. Classify each of the following as an element or a compound.

_____ a. benzene, C_6H_6

_____ b. aluminum, Al

_____ c. aspirin, $C_9H_8O_4$

_____ d. titanium, Ti

_____ e. acetylene, C_2H_2

_____ f. zinc, Zn

5. Explain why elements and compounds are pure substances.

6. Determine which of the following are pure substances and which are mixtures.

_____ a. salt water

_____ b. isopropyl alcohol, C_3H_8O

_____ c. mercury, Hg

_____ d. ammonia, NH_3

_____ e. an egg yolk

_____ f. honey

Holt Chemistry Section 2 Answers

Wenbin Ji

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Holt Chemistry Section 2 Answers:

The book delves into Holt Chemistry Section 2 Answers. Holt Chemistry Section 2 Answers is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Holt Chemistry Section 2 Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Holt Chemistry Section 2 Answers
 - Chapter 2: Essential Elements of Holt Chemistry Section 2 Answers
 - Chapter 3: Holt Chemistry Section 2 Answers in Everyday Life
 - Chapter 4: Holt Chemistry Section 2 Answers in Specific Contexts
 - Chapter 5: Conclusion
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