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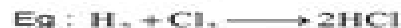
REDOX REACTIONS AND ELECTROCHEMISTRY

EARLIER CONCEPT OF OXIDATION

Oxidation is the process of addition of oxygen or any electronegative element or group or removal of hydrogen or any electropositive element or group.



Reduction is the process of addition of hydrogen or any electropositive element or group or removal of oxygen or any electronegative element or group.



ELECTRONIC CONCEPT

According to the electronic concept, oxidation involves loss of electrons or removal of electrons.



Here MnO_4^{2-} , Fe^{2+} and Mg are donating electrons (electron donor) and are called Reducing agents.

∴ Reducing agent is electron donor or we can say that reducing agents are getting oxidised. Similarly

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L Darling-Hammond



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