11. In the reaction: 4981, + 5O₂ + 4590 + 5H₂O₂ if 20 grams of NH, and 50 grams of O₂ are mixed, which is the limiting reactiont, and how many grams of NO can be produced? (Mose messec: NH, = 17 g/mot, O₂ = 22 g/mot, NO = 30 g/mot)

 For the reaction: 2At + 3H₂SO₂ + A₂SSO₂, + 3H₂, if 10 grams of At and 50 grams of H₂SO₂ are reacted, which reactest is limiting, and tow many grams of H₂ can be produced? (Molar masses: Al + 27 girest, H₂SO₂ + 95 g/mol, H₃ + 2 g/mol)

In the reaction: 2CuSO, * 2NayPO, * CuyPO₂), * 3NaySO₂, # 30 grains of CuSO, and 25 grains of NeyPO₂ are mixed, which is the limiting reactions, and how many grains of CuyPO₂, can be produced? (Note: masses: CuSO, * 160 g/mol. NeyPO₂ * 164 grand, CuyPO₂, * 382 grand)

For the reaction; 2KMnO₄ + SH₂C₂O₄ + 3H₂SO₄ + 2MnSO₄ + 1SCO₃ + 8H₂O + K₂SO₄.
If 20 grams of KMnO₄, 30 grams of H₂C₂O₄, and 25 grams of H₂SO₄ are reacted.
which reactant is limiting, and how many grams of CO₄ can be produced? (Motor measure: KMnO₄ + 158 g/mot, H₂C₂O₄ + 90 g/mot, H₂SO₄ + 96 g/mot. CO₄ + 44.
g/mot

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Michael Seilmaier

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