

Section 1: Energy

Example: Each example found by putting the correct sign in the space provided.

- a. The object is at rest. $\Delta K = 0$
- b. An object is moving downward and is slowing down. $\Delta K = \text{negative}$
- c. An object is moving downward and is speeding up. $\Delta K = \text{positive}$
- d. An object is moving upward and is slowing down. $\Delta K = \text{negative}$
- e. An object is moving upward and is speeding up. $\Delta K = \text{positive}$
- f. An object is moving downward and is slowing down. $\Delta K = \text{negative}$
- g. An object is moving downward and is speeding up. $\Delta K = \text{positive}$
- h. An object is moving upward and is slowing down. $\Delta K = \text{negative}$
- i. An object is moving upward and is speeding up. $\Delta K = \text{positive}$

Write the answer to the following questions in the space provided.

- a. What is the difference between kinetic and potential energy?
Kinetic energy is the energy of motion, while potential energy is the energy stored in an object due to its position or state.
- b. How is energy conserved in a closed system?
Energy is conserved in a closed system because the total energy remains constant. Energy can be transformed from one form to another, but it cannot be created or destroyed.

Holt Chemistry Concept Review Section 5 2 Answer Key

Tobias Bleicker

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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