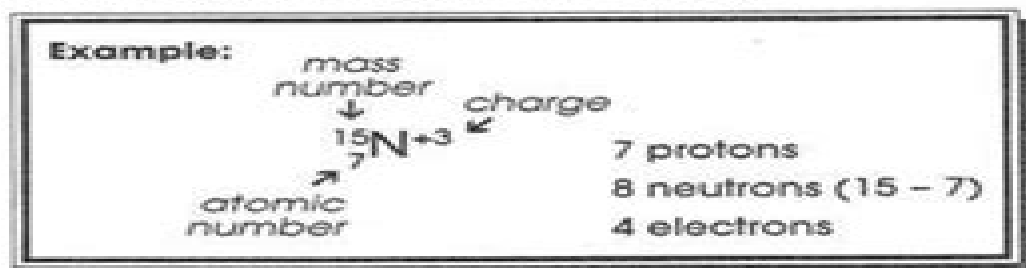


ATOMIC STRUCTURE

Name _____

An atom is made up of protons and neutrons (both found in the nucleus) and electrons (in the surrounding electron cloud). The atomic number is equal to the number of protons. The mass number is equal to the number of protons plus neutrons. In a neutral atom, the number of protons equals the number of electrons. The charge on an ion indicates an imbalance between protons and electrons. Too many electrons produces a negative charge, too few, a positive charge.

This structure can be written as part of a chemical symbol.



Complete the following chart.

Element/ Ion	Atomic Number	Atomic Mass	Mass Number	Protons	Neutrons	Electrons
H						
H ⁺						
¹² ₆ C						
⁷ ₃ Li ⁺						
³⁵ ₁₇ Cl ⁻						
³⁹ ₁₉ K						
²⁴ ₁₂ Mg ²⁺						
As ³⁻						
Ag						
Ag ⁺						
S ⁻²						
U						

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american samoa leone was the ancient capital of tutuila island leone was also where the samoan islands first missionary john
williams visited on october 18 1832 a monument in honor of williams has been erected in front of zion

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