

General Instructions:

- This question paper consists of 39 questions in 5 sections.
- All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- Section A consists of 20 objective type questions carrying 1 mark each.
- Section B consists of 6 very short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- Section C consists of 7 short answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- Section D consists of 3 long answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

Section-A

Select and write the most appropriate option out of the four options given for each of the questions 1 - 20.

1X20= 20M

Q.1) Given the reaction:



1

Which of the following statements is true?

- Zinc is reduced and copper is oxidised.
 - Zinc is oxidised and copper is reduced.
 - Both zinc and copper are oxidised.
 - Both zinc and copper are reduced.
- Q.2) The reaction between MnO_2 with HCl is depicted in the following diagram. It was observed that a gas with bleaching abilities was released.

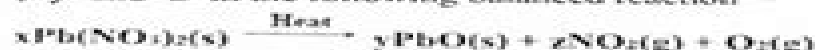
1



The chemical reaction between above mentioned reactants:

- displacement reaction
- combination reaction
- redox reaction
- decomposition reaction.

Q.3) Identify 'x', 'y' and 'z' in the following balanced reaction



1

(a) 2, 4, 2

(b) 2, 2, 4

(c) 2, 4, 4

(d) 4, 2, 2

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RS Peters



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