

Q3) Attempt any FOUR of the following : **[16]**

- a) Define the term cryoscopic constant. Derive thermodynamically the relation.

$$K_f = \frac{RT_0^2}{1000L_f}$$

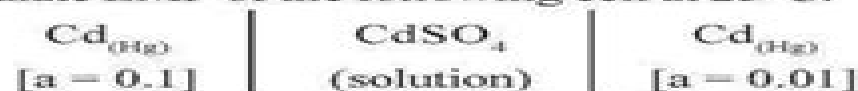
- b) Explain the effect of hydrogen bonding on the following physical properties.
- solubility and
 - Viscosity.
- c) Derive an expression for concentration cells with transference.
- d) A second order reaction where $a = b$, is 30% completed with in 500 seconds. How long will it take for the reaction to go to 70% completion?
- e) The osmotic pressure of a 5% by volume solution of a substance at 18°C is 6.64 atm. Calculate the molecular weight of the substance.
- f) Describe eutectic phase diagram for $\text{Zn} - \text{Cd}$ system.

Q4) Attempt any FOUR of the following : **[16]**

- a) What do you understand by 'geometrical isomerism'. Explain the use of E and Z nomenclature to describe geometrical isomers. Explain the stability of geometrical isomers.
- b) What is plane polarised light? How it is related to optical activity? Explain the use of polarimeter to measure optical activity.
- c) Explain the following types of organic reactions :
- Substitution reactions and
 - Oxidation and reduction reactions.
- d) Balance the following equation by oxidation number method :



- e) Calculate EMF of the following cell at 25°C .



- f) Findout the pH of a 0.002 M acetic acid solution if it is 2.3% ionised at this dilution.

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