- Q4) a) Attempt any three of the following: [12]
  - Describe any one method for the purification of colloids.
  - Explain the set of four quantum numbers.
  - iii) Explain positive and negative catalysis with suitable examples.
  - iv) Explain the general properties of colloids.
  - v) Describe Milikans oil drop method for determination of charge of the electron.
  - Attempt any one of the following:
    - What is hydrogen bond? What are the essential conditions to form hydrogen bond.

[4]

[6]

[10]

- What is tetrahedral hybridisation? Explain formation of CH<sub>4</sub> molecule on the basis of hybridisation.
- Q5) a) Attempt any two of the following:
  - Mention the various isotopes of hydrogen. Explain physical properties of deuterium.
  - Explain formation of N, molecule on the basis of overlap of atomic orbitals.
  - iii) Calculate the molarity of solution which contain 10 gms of K<sub>2</sub>CO<sub>3</sub> per 1000ml of water

(Given Molecular Weight of K,CO, = 138)

b) Attempt any two of the following:

i) Give postulates of Pauling's-slater theory.

- ii) 10 ml solution of NaOH containing 2 gms of alkali per litre is exactly neutralised by 15 ml of solution of H<sub>2</sub>SO<sub>4</sub> and 30 ml of HCl solution separately? Calculate the strength of the acid in gms per litre.
- iii) Give the assumptions of VSEPR theory.

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