

Concept Review

Section: Families of Elements

1. **Classify** each of the following elements as an alkali metal, alkaline-earth metal, transition metal, or semiconductor based on its position in the periodic table.

_____ **a.** rubidium, Rb

_____ **b.** silicon, Si

_____ **c.** silver, Ag

_____ **d.** barium, Ba

2. **Classify** each of the following elements as a halogen, noble gas, or other nonmetal based on its position in the periodic table.

_____ **a.** carbon, C

_____ **b.** chlorine, Cl

_____ **c.** radon, Rn

_____ **d.** phosphorus, P

3. **Predict** which of the following ions would be likely to form:

_____ **a.** Na^{2+}

_____ **d.** Br^-

_____ **b.** Cl^+

_____ **e.** Ne^-

_____ **c.** Ca^{2+}

_____ **f.** Ne^+

4. **Explain** why chlorine, Cl, is very reactive, whereas argon, Ar, is unreactive.

5. **Analyze** the following pairs of elements, and determine whether each pair has similar or different reactivities.

_____ **a.** potassium, K, and rubidium, Rb

_____ **b.** calcium, Ca, and barium, Ba

_____ **c.** sodium, Na, and chlorine, Cl

_____ **d.** helium, He, and krypton, Kr

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