

Naming Review Practice

Directions: On the left of each question, identify if the compound/molecule is Ionic (write an I) or Covalent (write a C). Then, either write the chemical name of the molecule (if given the formula) or the chemical formula of the molecule (if given the name)

- | | | |
|-----------------|---------------------------------|---------------------------------------|
| Covalent | 1. CS_3 | Carbon Trisulfide |
| Ionic | 2. TiO_2 | Titanium IV Oxide |
| Ionic | 3. CaCO_3 | Calcium Carbonate |
| Ionic | 4. $\text{Cu}(\text{NO}_2)_2$ | Copper II Nitrite |
| Ionic | 5. $\text{Sn}(\text{CN})_4$ | Tin IV Cyanide |
| Ionic | 6. Ag_2SO_3 | Silver Sulfite |
| Ionic | 7. NiSe_2 | Nickel IV Selenide |
| Covalent | 8. P_2O_5 | Triphosphorus Pentoxide |
| Ionic | 9. $\text{Sn}_3(\text{PO}_4)_2$ | Tin II Phosphate |
| Covalent | 10. N_7Cl_3 | Heptanitrogen Trichloride |
| Ionic | 11. Manganese (III) bromide | MnBr_3 |
| Covalent | 12. Sulfur dioxide | SO_2 |
| Ionic | 13. Tin (IV) sulfate | $\text{Sn}(\text{SO}_4)_2$ |
| Ionic | 14. Zinc hydroxide | $\text{Zn}(\text{OH})_2$ |
| Covalent | 15. Carbon tetrachloride | CCl_4 |
| Ionic | 16. Barium cyanide | $\text{Ba}(\text{CN})_2$ |
| Ionic | 17. Cobalt (VI) Chromate | $\text{Co}(\text{CrO}_4)_3$ |
| Ionic | 18. Lithium nitrite | LiNO_2 |
| Covalent | 19. Trisulfur hexafluoride | S_3F_6 |
| Ionic | 20. Aluminum oxalate | $\text{Al}_2(\text{C}_2\text{O}_4)_3$ |

Naming Review Guide Answer Key

Ying-Ying Zheng



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