Blackline Master 2.3g

Ionic Compounds: Names and Formulas Worksheet

- Write the formulas for the following compounds.
 - (a) sodium fluoride
 - (b) potassium chloride
 - (c) calcium oxide
 - (d) lithium bromide
 - (e) beryllium chloride
 - potassium sulfide
 - (g) magnesium nitride

- calcium phosphide
- (6)
- CD

- (m

- 10		
)	magnesium oxide	Mago
)	aluminum chloride	ATCL
()	sodium iodide	NaIS
)	potassium nitride	-K.N
m)	lithium sulfide	Suits Sex minus
1)	barium phosphide	Barp

- 2. Write the names for the following compounds.
 - chloride (a) LiCI
 - (b) MgS
 - (c) NaCl
 - (d) Al₂O₂
 - (e) CaS
 - Zn₂P₂

- Na₂0
- KBr
- (i) BeO
- (k) CaF₂ calcium
- BaBr₂ (0)
- (m) Na₂N
- (n) Al, S,

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T. R. Dickson

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Introduction to Chemistry, Laboratory Manual T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students Hazardous Materials Emergency Response ,1987

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