

NAMING IONIC COMPOUNDS

Name _____

Name the following compounds using the Stock Naming System.

and traditional
(where appropriate)

1. CaCO_3 calcium carbonate
2. KCl potassium chloride
3. $\overset{+2}{\text{Fe}}\text{SO}_4$ iron(II) sulfate ferrous sulfate
4. LiBr lithium bromide
5. MgCl_2 magnesium chloride
6. $\overset{+3}{\text{Fe}}\text{Cl}_3$ iron(III) chloride ferric chloride
7. $\text{Zn}_3(\text{PO}_4)_2$ zinc phosphate
8. NH_4NO_3 ammonium nitrate
9. $\text{Al}(\text{OH})_3$ aluminum hydroxide
10. $\overset{+1}{\text{Cu}}\text{C}_2\text{H}_3\text{O}_2$ copper(I) acetate cuprous acetate
11. $\overset{+2}{\text{Pb}}\text{SO}_3$ lead(II) sulfite plumbous sulfite
12. NaClO_3 sodium chlorate
13. CaC_2O_4 calcium oxalate
14. $\overset{+3}{\text{Fe}}_2\text{O}_3$ iron(III) oxide ferric oxide
15. $(\text{NH}_4)_3\text{PO}_4$ ammonium phosphate
16. NaHSO_4 sodium bisulfate / sodium hydrogen sulfate
17. Hg_2Cl_2 mercury(I) chloride mercurous chloride
18. $\text{Mg}(\text{NO}_2)_2$ magnesium nitrite
19. $\overset{+2}{\text{Cu}}\text{SO}_4$ copper(II) sulfate cupric sulfate
20. NaHCO_3 sodium bicarbonate / sodium hydrogen carbonate
21. $\overset{+3}{\text{Ni}}\text{Br}_2$ nickel(III) bromide
22. $\text{Be}(\text{NO}_3)_2$ beryllium nitrate
23. ZnSO_4 zinc sulfate
24. $\overset{+3}{\text{Au}}\text{Cl}_3$ gold(III) chloride
25. KMnO_4 potassium permanganate

Naming Compounds Containing Polyatomic Ions

Answers

RJ Shavelson



Naming Compounds Containing Polyatomic Ions Answers:

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