

Answer Key

60 + 4 ext

Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in...	Tastes...	Turns Litmus Paper...	Conducts Electricity?	Acid or Base
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	$\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	$\text{H}_3\text{PO}_4(\text{aq})$	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	$\text{HCl}(\text{aq})$	Stomach acid	Sour	Red	Yes	Acid
Citric acid	$\text{H}_3\text{C}_6\text{H}_5\text{O}_7(\text{aq})$	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	$\text{Mg}(\text{OH})_2(\text{aq})$	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	$\text{Al}(\text{OH})_3(\text{aq})$	Antacids	Bitter	Blue	Yes	Base
Barium hydroxide	$\text{Ba}(\text{OH})_2(\text{aq})$	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	$\text{NaOH}(\text{aq})$	Drain cleaner	POISON	Blue	Yes	Base

1. Refer to Model 1.
 - a. What is the common chemical name for Vitamin C?
Ascorbic acid
 - b. Is Vitamin C classified as an acid or a base?
Acid
2. Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
sour, turn litmus paper red, conduct electricity
3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
H at the beginning
4. Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.
bitter / poison, turn litmus paper blue, conduct electricity

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A Gutmann



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Table of Contents Naming Acids And Bases Tesccc Answer Key

1. Understanding the eBook Naming Acids And Bases Tesccc Answer Key
 - The Rise of Digital Reading Naming Acids And Bases Tesccc Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Acids And Bases Tesccc Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Acids And Bases Tesccc Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Acids And Bases Tesccc Answer Key
 - Personalized Recommendations
 - Naming Acids And Bases Tesccc Answer Key User Reviews and Ratings
 - Naming Acids And Bases Tesccc Answer Key and Bestseller Lists

5. Accessing Naming Acids And Bases Tesccc Answer Key Free and Paid eBooks
 - Naming Acids And Bases Tesccc Answer Key Public Domain eBooks
 - Naming Acids And Bases Tesccc Answer Key eBook Subscription Services
 - Naming Acids And Bases Tesccc Answer Key Budget-Friendly Options
6. Navigating Naming Acids And Bases Tesccc Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Acids And Bases Tesccc Answer Key Compatibility with Devices
 - Naming Acids And Bases Tesccc Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Acids And Bases Tesccc Answer Key
 - Highlighting and Note-Taking Naming Acids And Bases Tesccc Answer Key
 - Interactive Elements Naming Acids And Bases Tesccc Answer Key
8. Staying Engaged with Naming Acids And Bases Tesccc Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Acids And Bases Tesccc Answer Key
9. Balancing eBooks and Physical Books Naming Acids And Bases Tesccc Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Acids And Bases Tesccc Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Naming Acids And Bases Tesccc Answer Key
 - Setting Reading Goals Naming Acids And Bases Tesccc Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Naming Acids And Bases Tesccc Answer Key
 - Fact-Checking eBook Content of Naming Acids And Bases Tesccc Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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