Name:		Date:	Period:	
	SOLUBI	LITY WOI	RKSHEET	
Use the solubility curv	ves on the attached p	sage to answer th	e following questions.	
I. These solubility cur	wes are based on the s	olubility (in gran	is) of various salts	
dissolving in	grams of water.	In general for so	slids, as	
temperature increases,	the solubility of the sa	allt		
2. Most and least solub	ble at different temper	ratures:		
a. Which salt is beg	ast soluble at 20°C?_	753		
b. Which salt is me	ost soluble at 10°C7_			
c. For which salt d	loes the solubility char	nge the least?		
3. How many grams of following solutions?	f each solute below is	needed to saturat	e 100g of water at 80 °C t	o prepare the
a. KNO ₁				
b. KCI				
c. NaCl				
4. How many grams of	f each solute below as	e needed to make	a saturated solution at	50 °C?
a. KCIO ₃				
b. NaNO3				
c. KNO _{3.}				
5. At what temperature	e are the following so	lutes equally solu	ble in 100g of water?	
a. NaNO ₃ and KNO	Ο,			
b. KNO ₃ and NaCl	6			
c. NaCl and KCl				
6. The quantity of solu saturated, unsaturate			ome solutions is listed bel	ow. Is the soluti

Solute	Lemperature	
a. 103 g KNO ₃	60°C	150
b. 40 g KCI	40°C	

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Terry C. Jones

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