

- **Thermodynamics** is the study of energy and its transformations.
 - **First Law of Thermodynamics**: Energy cannot be created or destroyed, only converted from one form to another.

$$\Delta U = Q - W$$

 where ΔU is the change in internal energy, Q is the heat added to the system, and W is the work done by the system.



- **Second Law of Thermodynamics**: The entropy of an isolated system never decreases.
 - **Entropy** is a measure of the disorder or randomness of a system.
 - **Heat Engines**: Devices that convert heat into work. The efficiency of a heat engine is given by:

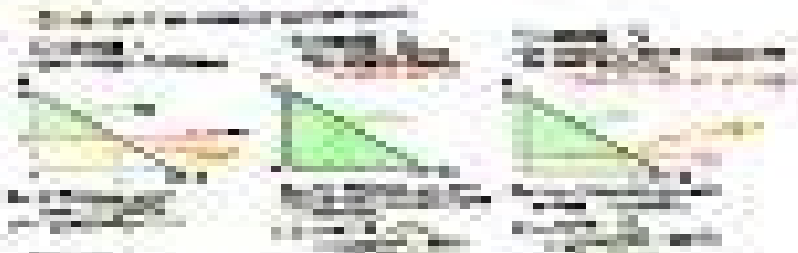
$$\eta = \frac{W}{Q_{in}} = 1 - \frac{Q_{out}}{Q_{in}}$$



- **Third Law of Thermodynamics**: The entropy of a perfect crystal at absolute zero is zero.
 - **Heat Pumps and Refrigerators**: Devices that transfer heat from a cold reservoir to a hot reservoir. The coefficient of performance (COP) for a refrigerator is:

$$COP = \frac{Q_{out}}{W} = \frac{1}{\frac{1}{\eta} - 1}$$

- **Statistical Mechanics**: The study of the microscopic properties of matter in terms of the statistical behavior of a large number of particles.



- **Maxwell's Equations**: A set of four partial differential equations that describe the behavior of electric and magnetic fields.

$$\nabla \cdot \mathbf{E} = \frac{\rho}{\epsilon_0}$$

$$\nabla \cdot \mathbf{B} = 0$$

$$\nabla \times \mathbf{E} = -\frac{\partial \mathbf{B}}{\partial t}$$

$$\nabla \times \mathbf{B} = \mu_0 \mathbf{J} + \mu_0 \epsilon_0 \frac{\partial \mathbf{E}}{\partial t}$$

- **Electromagnetic Induction**: The process by which a changing magnetic field induces an electric field.

$$\mathcal{E} = -\frac{d\Phi_B}{dt}$$

- **Electromagnetic Waves**: Waves that propagate through space, carrying energy and momentum.

$$E = cB$$


- **Quantum Mechanics**: The study of the behavior of matter and energy at the atomic and subatomic scales.

$$E = hf$$


Microeconomics Final Study Guide

O. Ray Whittington



Microeconomics Final Study Guide:

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