



Nuclear Equations

(a) Disintegration of radium-236 by emission of an alpha particle (4 He).

$$^{236}_{55}$$
Ra \longrightarrow $^{232}_{56}$ Rn + $^{4}_{2}$ He

Mass Nos: Reactants = 236; Products = 232 + 4 = 236

Atomic Nos: Reactants = 88; Products = 86 + 2 = 88

(b) Disintegration of phosphorus-32 by emission of a beta particle $\begin{pmatrix} 0 \\ -1 \end{pmatrix} e$,

$$^{32}_{15}P \longrightarrow ^{0}_{-1}e + ^{32}_{16}S$$

Mass Nos: Reactants = 32: Products: 0 + 32 = 32

Atomic Nos: Reactants = 15; Products: 16-1=15

(c) Fission of argon-40 by bombardment with a proton H.

Mass Nos: Reactants = 40 + 1 = 41; Products = 40 + 1 = 41

Atomic Nos: Reactants = 18 + 1 = 19; Products = 19 + 0 = 19

Mcq Nuclear Equation

JA Banks

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