

(a)

Reactants	Products
H, Cl and HCl	2H, O ₂ and H ₂ O

(b)

Reactants	Products
HCl and 2H ₂ O	H, Cl, 2H and O ₂

(c)

Reactants	Products
H, Cl, 2H and O ₂	HCl and 2H ₂ O

(d)

Reactants	Products
2H, O ₂ and H ₂ O	H, Cl and HCl

Correct Answer: Option C

Q4) A student performs an experiment to form aluminium chloride from aluminium and chlorine. Which options give the chemical equation of the reaction?

- (a) $\text{Al} + \text{Cl}_2 \rightarrow \text{AlCl}_2$
 (b) $2\text{Al} + \text{Cl}_2 \rightarrow 2\text{AlCl}$
 (c) $2\text{Al} + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3$
 (d) $3\text{Al} + 3\text{Cl}_2 \rightarrow 3\text{AlCl}_3$

Correct Answer: Option C

Q5. Give the ratio in which hydrogen and oxygen are present in water by volume.

- (a) 1:2
 (b) 1:1
 (c) 2:1

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Lingsheng Yao



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1. Understanding the eBook Mcq Bank On Chemical Reaction And Equation
 - The Rise of Digital Reading Mcq Bank On Chemical Reaction And Equation
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcq Bank On Chemical Reaction And Equation
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcq Bank On Chemical Reaction And Equation
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcq Bank On Chemical Reaction And Equation
 - Personalized Recommendations
 - Mcq Bank On Chemical Reaction And Equation User Reviews and Ratings
 - Mcq Bank On Chemical Reaction And Equation and Bestseller Lists
5. Accessing Mcq Bank On Chemical Reaction And Equation Free and Paid eBooks
 - Mcq Bank On Chemical Reaction And Equation Public Domain eBooks
 - Mcq Bank On Chemical Reaction And Equation eBook Subscription Services
 - Mcq Bank On Chemical Reaction And Equation Budget-Friendly Options

6. Navigating Mcq Bank On Chemical Reaction And Equation eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcq Bank On Chemical Reaction And Equation Compatibility with Devices
 - Mcq Bank On Chemical Reaction And Equation Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcq Bank On Chemical Reaction And Equation
 - Highlighting and Note-Taking Mcq Bank On Chemical Reaction And Equation
 - Interactive Elements Mcq Bank On Chemical Reaction And Equation
8. Staying Engaged with Mcq Bank On Chemical Reaction And Equation
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcq Bank On Chemical Reaction And Equation
9. Balancing eBooks and Physical Books Mcq Bank On Chemical Reaction And Equation
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcq Bank On Chemical Reaction And Equation
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mcq Bank On Chemical Reaction And Equation
 - Setting Reading Goals Mcq Bank On Chemical Reaction And Equation
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcq Bank On Chemical Reaction And Equation
 - Fact-Checking eBook Content of Mcq Bank On Chemical Reaction And Equation
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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