বিষয় কোত :	1	7	6
Ø6:			7

হচনিৰ্বাচনি অভীকা

TO PROPERTY

ক্রবাদ পর

(২০২৩ সালের সিলেবাস অনুযায়ী)

সময়—২৫ মিনিট

বিলেৰ দুইবা । সৰবৰাহকৃত বছনিৰ্বাচনি অভীকাৰ উত্তৰপত্তে প্ৰস্তেই ক্ৰমিক নম্বৰে বিপত্তীতে প্ৰকল্প বৰ্ণসম্পদত বৃত্তসমূহ হতে স্তিক/সৰ্বোধ্যক্ট উত্তৰে বৃত্ততি বল পছেন্ট কলম ছাৱা সম্পূৰ্ণ তথাট কৰে । প্ৰতিটি প্ৰস্তেৱ মান ১ ।]

অপ্লশতত কোনো প্ৰকাৰ দাশ/চিফ দেৱা যাবে দা।

- চিকিৎসা বিজ্ঞানে কিঞ্জিতখেরাপিতে কোনটি ব্যবহার করা হয়?
 - (P) X-ray

সিলেট বোর্ড

- B IR
- (9) MRI
- (M) UV

উদ্দীপক থেকে ২ এবং ৩ নং প্রপ্লের উত্তর দাও :

মৌল	বহিন্তে জনের ইলেবট্রন বিন্যাস			
A	(n-1)s ¹	Maked		
В	ns²np⁴	L decision		
С	(n+1)s ² (n+1)p ⁴	मान 2		

A, B এবং C প্রচলিত থৌলের প্রতীক বর।

- ২। 🗛 ৪ ৩ 🗛 C ভৌত অবস্থার ভিন্নতার ভারণ---
 - (ক) আছনিক বছন
 - (খ) সমযোগী বন্ধল
 - হাইছেলেন বছন
 - (খ) সন্নিবেশ বন্ধন
- ৩। A2B খৌগের কেতর—
 - ে sp³ সংকলণ ফটে
 - AL বছন কোপ 104-5°
 - iii. (Lp Lp) विकर्षन विनामान

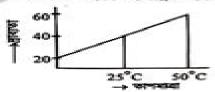
নিচের কোনটি সঠিক?

- (4) 14 H
- (4) II 4 III
- ৪। নিজের মৌলভলার ১ম আছনীকরণ বিভারের সঞ্জিক ক্রম কোনটিঃ
 - (*) Be>B>N>O
 - N > O > Be > B
 - (*f) O>N>B>Be (*f) B>Be>N>O

- ৫। গোটনের প্রকৃত ভর কত।
 - (*) 1-60×10⁻²⁴g
 - (4) 1-66×10⁻²⁴g
 - 9 1-673×10⁻²⁴g
 - (v) 1-675×10⁻²⁴g
- **৯। কোনটি প্রশন্ন অক্সাইড**া
 - (4) N₂O₃
- (4) NO³

12

- । S-ব্ৰক খৌলের সংখ্যা ককটিঃ
 - (**) 7 (
- **9** 14 (%) 16
- শরমাপুর ৩য় শক্তিকরে ফুর্লনরত ইলেকট্রনের কৌশিক ভরবেশ এর মান কোনটিঃ
- 3h 2×
- (四) 经
- া কোনটি উভহবৰী অপ্লাইত।
 - (1) KO₂
- (4) Na₂O₂ (4) Fe₂O₃
- উধীলক হতে ১০ এবং ১১ নং প্রপ্রের উত্তর দাও :



- ে। 200g NH₄CI এর সম্পৃত প্রকাকে 50°C থেকে 25°C তালধারার ইত্তে করলে কর প্রাথ NH₄CI কেলালিত হবে।
 - (m) 75

- (4) 50 (4) 90
 - /0 [भद्र भृक्षेत्र शुक्रेत

Hsc Chemistry Exam Mcq Question Out

Marilyn Schell, Margaret Hogan

Hsc Chemistry Exam Mcq Question Out:

Cambridge Checkpoints HSC Chemistry 2017-19 Maureen Slade, Roger Slade, 2016-06-29 Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every sin gle dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured. This guide comes in a brand new format that makes even better use of your study time up to date covera ge of the core topics plus 3 Option topics Industrial Chemistry Shipwr ecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to s ection the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second h and Investigations and Problem Solving %this way you will be ab le to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the E xcel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test you recall of knowledge in each chapter HSC type quest ions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way yo u can test yourself on HSC type questions under HSC type time pressure an examiner maximiser feature ticks to show the mark distribut ion and answers to all HSC type questions %all you need to answ er HSC type questions two sample HSC papers with an examiner m aximiser feature plus answers %not one but two up to date sample papers the Excel syllabus summary notes a detachable secti on at the end of the guide where every single dot point of each chapter is summarised for you% a comprehensive and compact summary of the whole course in 32 pages Excel Revise in a Month HSC Chemistry Will Marchment, 2007 Excel HSC Chemistry C. M. Roebuck, 2003 Preliminary Senior Science Jennifer Hill, 2011 This comprehensive study guide covers the complete HSC Preliminary Senior Science course and has been specifically created to maximise exam's uccess This guide has been designed to meet all study needs providing up to date information in an easy to use format The sample HSC Exam has been updated for the new format Excel HSC Preliminary Senior Science contains an introductory section including how to use the book and an explanation of the new course helpful study and exam techniques comprehensive coverage of the entir e Preliminary and HSC courses hundreds of diagrams to aid under standing icons and boxes to highlight key concepts and assessme nt skills including laboratory and field work checklists of key terms end of chapter revision questions with fully explained a nswers a trial HSC style exam with answers and explanations a glossary of key terms useful websites highlighted throu ghout Macauarie

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Australian secondary education looks beyond clich s about excellence to analyse the historically specific nature of the scholarly ideal against which successive generations of Australian students have been judged Drawing on a wealth of strikingly original research Richard Teese offers a penetrating analysis of Victorian secondary schooling in the half century after World War Two This was a era in which higher secondary schooling ceased to be the preserve of an elite and emerged as a system of mass education It was also a period marked by successive waves of reform in curriculum and assessment Yet at the end of it all Australians have been left with a sharply polarised system of schooling in which the most economically vulnerable populations of young people are also those most at risk of educational failure This book asks the hard questions Are our systems of secondary teaching and the expectations they place on students anachronistic in an age of mass education How far is the curriculum itself responsible for the manifest disparities in achievement between sectors and regions and between boys and girls What has been the universities role in the process of reform and counter reform And what have all these upheavals implied for the practice of teaching The Rhodesia Science News ,1975 **HSC Chemistry Marilyn** Schell, Margaret Hogan, 2007-09-01 In this book you will find nearly 600 typical multiple choice examination questions nearly 600 extension questions plus summaries for the Year 12 Chemistry course Answers to all questions are provided Questions follow the dot points in the Board of Studies syllabus and focus on the three core topics plus the option topic Shipwrecks Corrosion and Conservation Information Malaysia, 1980

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